

TO OPEN...
CUT HERE!



10638 ferti•lome® Chelated Iron EDDHA 6% Water Soluble

ferti•lome®

**For Soils With
A pH of 7.0 and
Above**

Chelated Iron **EDDHA 6%** **Water Soluble**

- Water Soluble Chelated Iron
- For Use On Alkaline Soils To Prevent Or Correct Chlorosis
- For Use On Trees, Shrubs, Ornamentals, Flowers, Vegetables, Plants In Containers And Lawns

KEEP OUT OF REACH OF CHILDREN



NET WEIGHT 4 LBS. (1.8 KG)

10638

ferti•lome®

Chelated Iron

EDDHA 6%

Water Soluble

ferti•lome® Chelated Iron EDDHA 6% Water Soluble is a water-soluble source of iron that is readily available to plants. It can be used to prevent or correct chlorosis on calcareous soils or soils with a pH above 7.0.

How to use ferti•lome® Chelated Iron EDDHA 6% Water Soluble

Soil Application

Apply in the Spring as a preventative or as needed in-season to correct chlorosis. Apply directly to the soil and water in, or mix with enough water to cover the area and apply as a drench, then water in. Apply early morning or late afternoon to reduce chances of burning. Always water after application to move the product into the rootzone.

Trees (Including Fruit, Nut and Citrus)

Use 1 to 4 Tbsp per one inch of trunk diameter (measured at 3 to 4 feet) and water immediately. Apply at the drip line.

Ornamentals and Shrubs

Use 1 to 1½ tsp per plant and water immediately. Apply evenly under the plant.

Flowers & Vegetables

Use ½ to ¾ tsp per plant or 2 to 4 Tbsp per 100 sq. ft. or sidedress 2 to 3 oz. per 100 ft. row and water immediately.

Plants in Containers

For an 8 inch pot use ¼ tsp, for a 12 inch pot use ½ tsp, apply and water immediately.

Lawns

Use 2 to 4 oz. per gallon of water per 500 sq. ft. and water in immediately.

GUARANTEED ANALYSIS

Iron (Fe)	6%
6% Chelated Iron (Fe)	
Derived From: Ferric ethylenediamine (FeEDDHA)	F370



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Visit Us At: www.fertilome.com
Product Questions? 855-270-4776

10638-0120-REG3



Information regarding the contents and levels of metals in this product is available on the internet at <http://www.aapfco.org/metals.html>